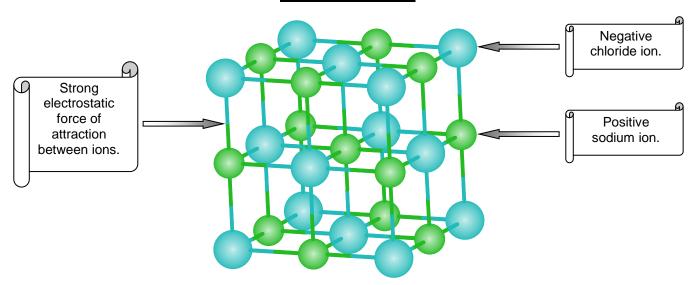


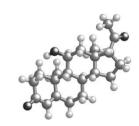
# The Structures and Properties of Materials <u>lonic Compounds</u>



**Diagram 1:** The lattice structure of sodium chloride.

| 1. | What type of bonding is present in sodium chloride?   |
|----|---|
| 2. | What best describes the structure of sodium chloride? |
| 3. | What are the properties of sodium chloride?           |
|    | Property:   |
|    | → Explanation:  |
|    | • Property:   |
|    | → Explanation:  |
|    | Property:   |
|    | → Explanation:  |
|    | • Property:   |
|    | → Explanation:  |





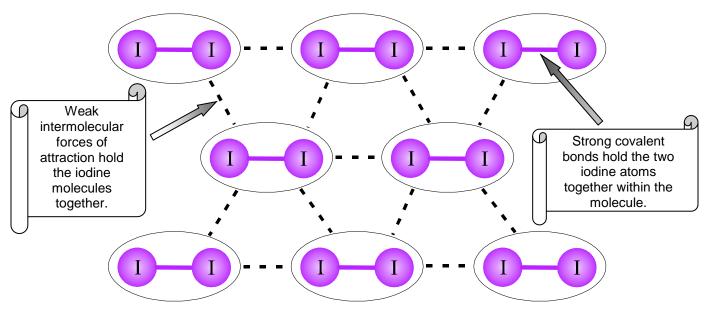
## Chem!stry class:

| Class: |  |  |  |
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Name: ..... ( )

Date: ..... / ..... / .....

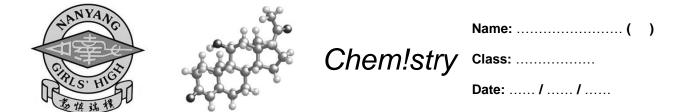
## **Structures and Properties of Materials Simple Covalent Elements and Compounds**



**Diagram 1:** This shows the bonding in a sample of solid iodine.

| 1. | What type of bonding is present in iodine?   |
|----|--|
| 2. | What best describes the structure of iodine? |
| 3. | What are the properties of iodine?           |
|    | Property:                                    |
|    | $\rightarrow$ Explanation:                   |
|    | Property:                                    |
|    | → Explanation:                               |
|    | Property:                                    |
|    | → Explanation:                               |
|    | Property:                                    |

→ Explanation: .....



### **Diamond**

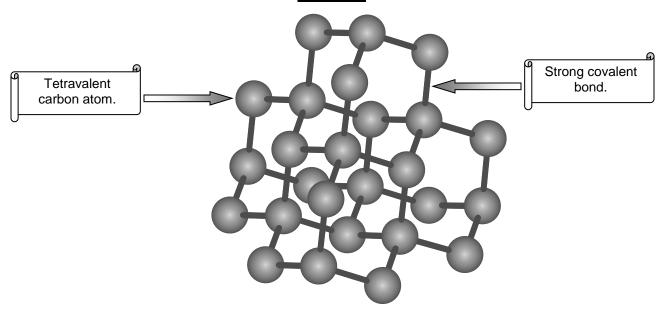
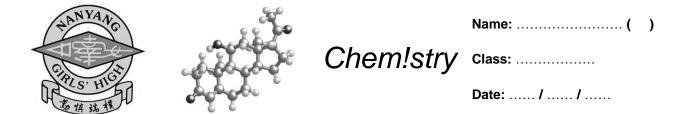
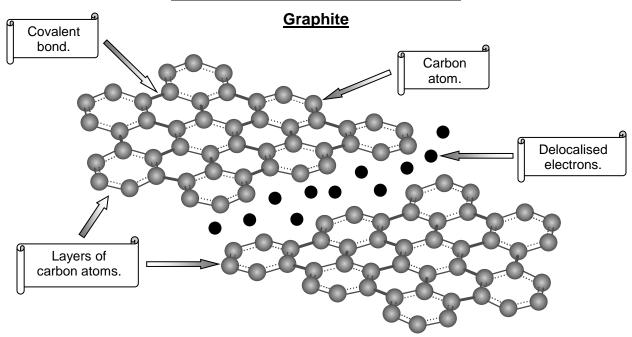


Diagram 1: This shows the structure of diamond, an allotrope of carbon.

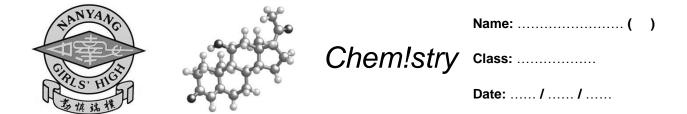
|    | , ,   |
|----|---|
| 1. | What type of bonding is present in diamond?   |
| 2. | What best describes the structure of diamond? |
| 3. | What are the properties of diamond?           |
|    | Property:                                     |
|    | → Explanation:                                |
|    | Property:                                     |
|    | → Explanation:                                |
|    | Property:                                     |
|    | → Explanation:                                |
|    | Property:                                     |
|    | → Explanation:                                |



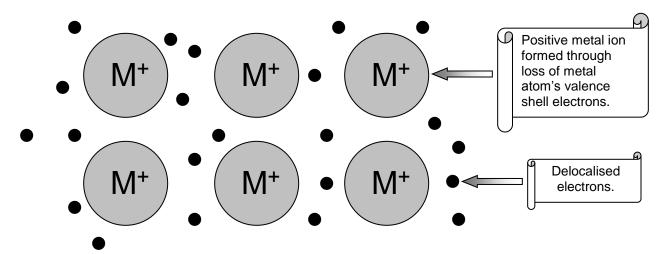


**Diagram 1:** This shows the structure of graphite, an *allotrope* of carbon.

| 1. | What type of bonding is present in graphite?            |
|----|---|
| 2. | What best describes the structure of graphite?          |
| 3. | What are the properties of graphite?  • Property:       |
|    | → Explanation:  |
|    | ◆ Property:      → Explanation:                         |
|    | Property:   |
|    | <ul> <li>→ Explanation:</li> <li>• Property:</li> </ul> |
|    | → Explanation   |

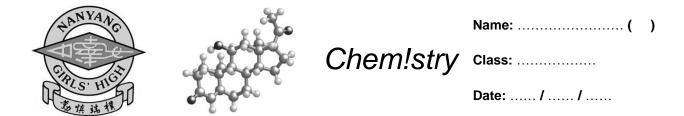


### **Metals**



**Diagram 1:** This shows the lattice structure of a metal.

| 1. | What type of bonding is present in a metal?   |
|----|---|
| 2. | What best describes the structure of a metal? |
| 3. | What are the properties of a metal?           |
|    | Property:                                     |
|    | → Explanation:                                |
|    | • Property:                                   |
|    | → Explanation:                                |
|    | • Property:                                   |
|    | → Explanation:                                |
|    | • Property:                                   |
|    | → Explanation:                                |



#### **Polymers**

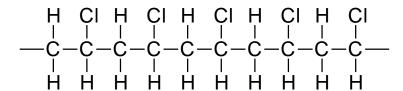


Diagram 1: This shows the structural formula of the polymer poly(chloroethene). The common name of this polymer is polyvinylchloride, or PVC for short.

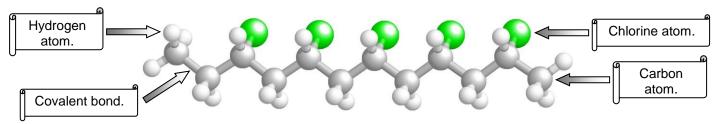


Diagram 2: This shows the "ball and stick" structure of the polymer poly(chloroethene).

1. What type of bonding is present in a polymer? **2.** What best describes the structure of a polymer? **3.** What are the properties of a polymer? • Property: ..... → Explanation: ..... • Property: ..... → Explanation: ..... • Property: ..... → Explanation: ..... Property: → Explanation: .....